



0040-4020(94)E0041-Q

Dilithiated 2-(Chloromethyl)-3-tosylpropene: A New γ -Chlorinated Allyl Sulfone Dianion in Organic Synthesis

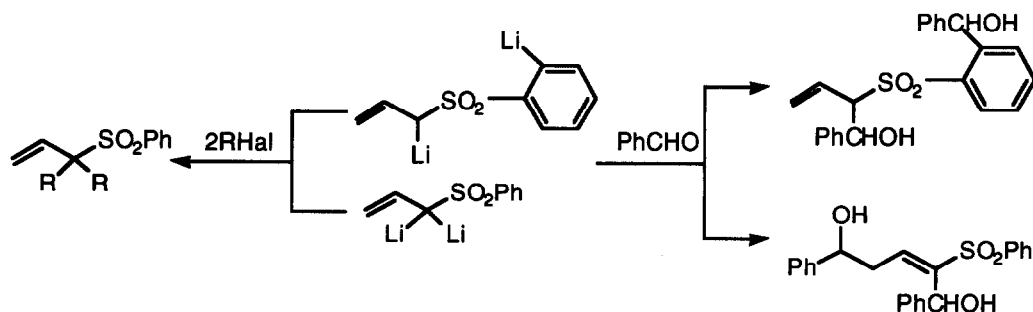
Carmen Nájera* and José M. Sansano

Departamento de Química Orgánica, Facultad de Ciencias, Universidad de Alicante, Apdo. 99, 03080 Alicante, Spain

Abstract: Dilithiation of 2-(chloromethyl)-3-tosylpropene (**1**) with *n*-butyllithium at -90°C in the presence of DMPU affords the allylic dianion **2** which reacts with deuterium oxide or very reactive alkylating agents to give α,α -disubstituted products **3** or **4**, respectively. Monoalkylation at the α -position followed by cyclopropanation reaction takes place with *n*-alkyl bromides to provide tosylated methylenecyclopropanes **5**. The reaction of dianion **2** with aldehydes or electrophilic olefins occurs at the α -position followed by the corresponding annulation process obtaining tosylated 2,5-dihydrofurans **6** or cyclopentenes **8**, respectively. When ketones are used as electrophiles, dianion **2** reacts at the γ -position yielding 3-(tosylmethylene)tetrahydrofurans **7**. Reductive desulfonylation of compounds **5-8** with sodium amalgam or samarium(II) iodide to afford the corresponding desulfonylated derivatives **22-26** has been also studied.

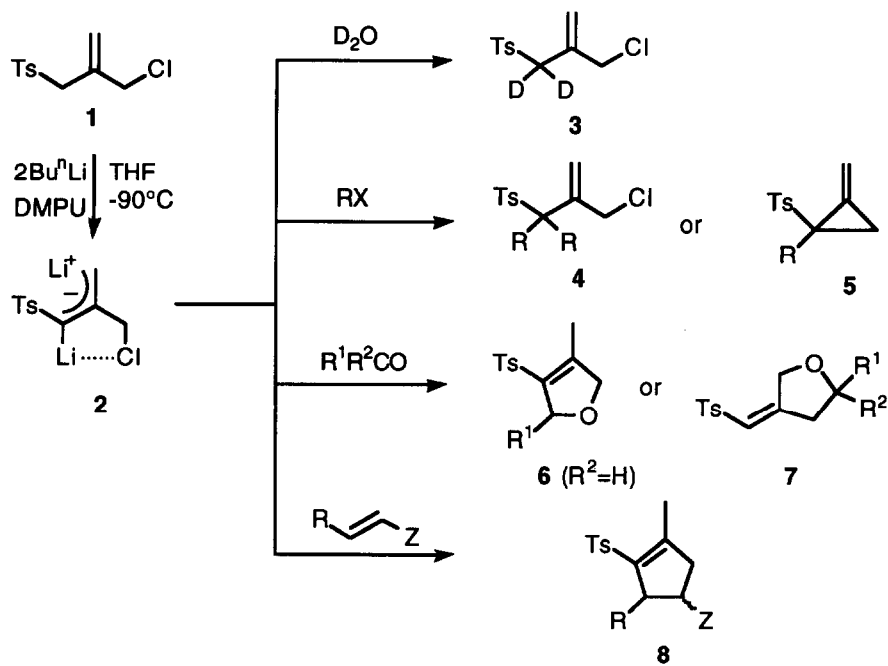
INTRODUCTION

The utility of sulfones as building blocks is mainly based in their ability to stabilize carbanions. Specially useful are monolithio derivatives of allylic sulfones which have been used in regioselective carbon-carbon bond formation mainly in the synthesis of various isoprenoid structures due to their general tendency to undergo α - rather than γ -alkylation¹. Dilithiated allyl phenyl sulfone is the only dilithio derivative described², which has been used in the synthesis of carba-prostacyclins^{2b}. The dilithiation of allyl phenyl sulfone at -50°C takes place in the α - and *ortho*-positions, however, double alkylation reaction of this dianion occurs only at the α -position, in contrast with benzaldehyde reacts at the α - and *ortho*-positions. At 50°C the α,\textit{ortho} -dianion is converted into the thermodynamically more stable α,α -dilithio intermediate which suffers dialkylation reaction also at the α -position and reacts with benzaldehyde at the α and γ -positions of the allyl moiety^{2a} (Scheme 1). We have studied the dilithiation of 2-(chloromethyl)-3-tosylpropene (**1**), a γ -chlorofunctionalized allyl sulfone with isobutene structure³, and the reactivity and synthetic applications of these new allyl sulfone dianion as bifunctional conjunctive reagent⁴.



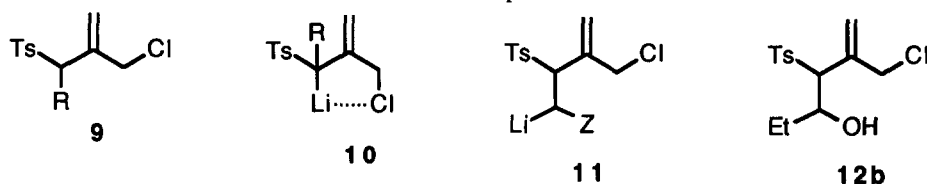
RESULTS AND DISCUSSION

When 2-(chloromethyl)-3-tosylpropene (**1**) was treated with two equivalents of *n*-butyllithium⁵ in THF at -90°C in the presence of *N,N'*-dimethylpropyleneurea (DMPU, two equivalents)⁷ over 15 min the corresponding dianion **2** was formed. This intermediate is stable at -90°C for one hour, decomposes at higher temperatures to give an intractable mixture of products and reacts with different electrophiles regioselectively depending on the kind of electrophilic reagent (Scheme 2). The structure for dianion **2** has been proposed according to the chemical behaviour of this intermediate in comparison with the monolithium derivative.^{3b,6} The treatment of intermediate **2**



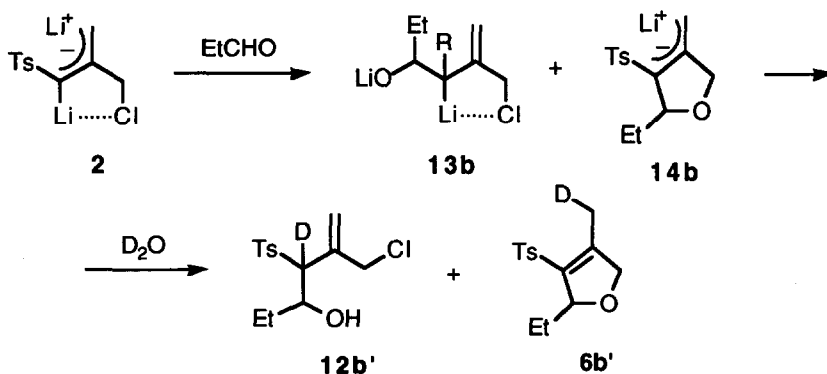
Scheme 2

with deuterium oxide afforded the α,α -dideuterated compound **3** in 72% yield and 90% of deuterium incorporation corresponding to two atoms of deuterium (^1H and ^{13}C NMR) (Scheme 2). We never observe deuterolysis at the *ortho*-position in the aromatic ring as in the case of allyl phenyl sulfone². By reaction of dianion **2** with very reactive alkylating agents, α,α -dialkylated **4** and or monoalkylated **9** compounds were obtained (Scheme 2 and Table 1). In the case of α -bromoacetonitrile and ethyl α -bromoacetate monoalkylated products **9f** and **9g**, respectively, were isolated, probably due to the transformation of the organolithium compound **10** into intermediates **11**. When less reactive alkyl bromides or benzyl chloride were allowed to react with dianion **2** only tosylated methylenecyclopropanes **5^b** were obtained, even with a large excess of electrophile (Scheme 2 and Table 1). The monoalkylated monoanion **10** suffers γ -elimination⁹, better than the second alkylation reaction between -30°C and room temperature to give compounds **5**. This cyclization was never observed when dianion **2** was allowed to warm till room temperature.



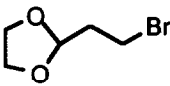
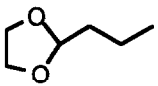
The reaction of intermediate **2** with aldehydes took place at the α -position to give, after intramolecular etherification, tosylated 2,5-dihydrofurans **6¹⁰** (Scheme 2 and Table 2). However, with ketones it reacted also regioselectively at the less hindered γ -position to afford tosylated methylenetetrahydrofurans **7**, whose stereochemistry was assigned according to the positive *nOe* effect between the vinylic proton and the methylene group in β respect to the oxygen atom (Scheme 2 and Table 2).

The reaction of dianion **2** with aldehydes took place between -90°C and room temperature. When the reaction with propanal was quenched with water at different temperatures, mixtures of compounds **6b** and **12b** (as mixture *erythro/threo*:1/2, deduced by ^1H NMR¹¹) were obtained. If deuterium oxide was added at -40°C a mixture of deuterated compounds **6b'** and **12b'** (*ca.* 1:3 molar ratio; 32% yield) was isolated (Scheme 3). It



Scheme 3

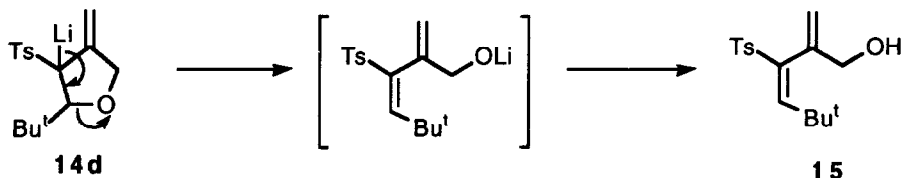
Table 1. Alkylation Reactions of Dianion **2**.

Electrophile ^a	Product			
	No.	R	yield (%) ^b	mp. (°C) ^c or <i>R_f</i> ^d
Me ₂ SO ₄	4a, 9a	Me	47, 31 ^e	0. 51, 0.46
CH ₂ =CHCH ₂ Br	4b, 9b	CH ₂ =CHCH ₂	46, 38 ^e	0.62, 0.53 ^f
HC≡CCH ₂ Br	4c	HC≡CCH ₂	62	141-142
PhCH ₂ Br	4d, 9d	PhCH ₂	38, 31 ^e	0.60, 0.50
I(CH ₂) ₃ Is	4e	-h	20	0.56
BrCH ₂ CN	9f	CH ₂ CN	52	130-131
BrCH ₂ CO ₂ Et	9g	CH ₂ CO ₂ Et	63	101-102
EtBr	5a	Et	54	0.50
n-BuBr	5b	n-Bu	45	0.53
Me ₃ SiCH ₂ Br	5c	Me ₃ SiCH ₂	51	0.52
THPOCH ₂ CH ₂ Br	5d	THPOCH ₂ CH ₂	33	0.30
EtOCH(Me)OCH ₂ CH ₂ Br	5e	HOCH ₂ CH ₂ ⁱ	60	0.42j
	5f		52	0.15
PhCH ₂ Cl	5g	PhCH ₂	33	59-60

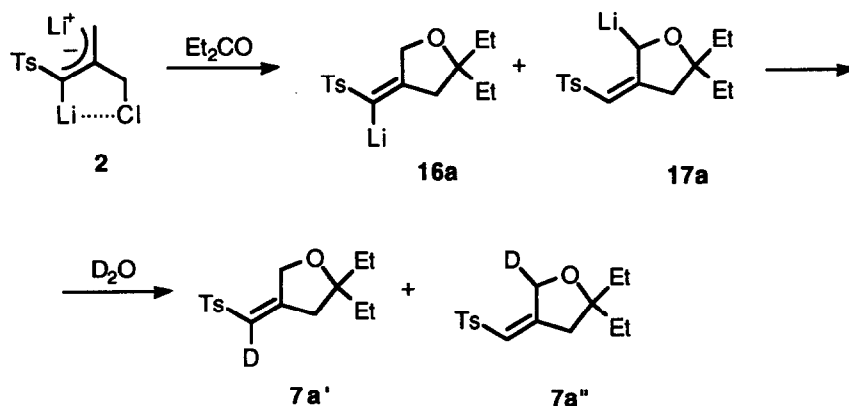
^a Two equivalents were added. ^b Isolated yield based on compound **1**, after column chromatography on silica gel. ^c Hexane/ether. ^d Hexane/ether: 1/1. ^e Compounds **4** and **9** were separated by column chromatography. ^f Lit.⁶ oil. ^g 1.2 Equivalents. ^h R-R=(CH₂)₃. ⁱ Deprotection of the ketal took place during chromatography. ^j Ether.

means that the precursor of compounds **12** is the β -oxido organolithium compound **13b**, which cyclizes slowly to give the less stable intermediate **14b**¹² precursor of dihydrofuran **6b** (Scheme 3).

Only in the case of the reaction of dianion **2** with pivalaldehyde the 2-tosyl-1,3-diene **15**¹³ was exclusively obtained due to the β -elimination reaction of the corresponding β -alkoxido organolithium compound **14d** (Scheme 4). The *E*-configuration of compound **15** was determined from ¹H NMR data for vinylic protons in β -position respect to the sulfone group in such type of sulfonyl dienes¹⁴.

**Scheme 4**

The reaction of dianion **2** with ketones takes place at temperatures ranging between -90° and -15°C to lead to 3-(tosylmethylene)tetrahydrofurans **7**. The addition to the carbonyl group and the cyclization processes occur simultaneously since when the reaction of intermediate **2** with 3-pentanone was quenched with water at different temperatures only compound **7a** and starting chlorosulfone **1** were obtained. When this reaction was quenched with deuterium oxide at -40°C compounds **7a'** and **7a''** (ca. 3:7 molar ratio; 47% yield) were obtained. At -15°C compound **7a''** was the only one isolated in 52% yield. Monoanions **16a** and **17a** seem to be in this case the intermediates, monolithium compound **17a** being the most thermodynamically stable one (Scheme 5).



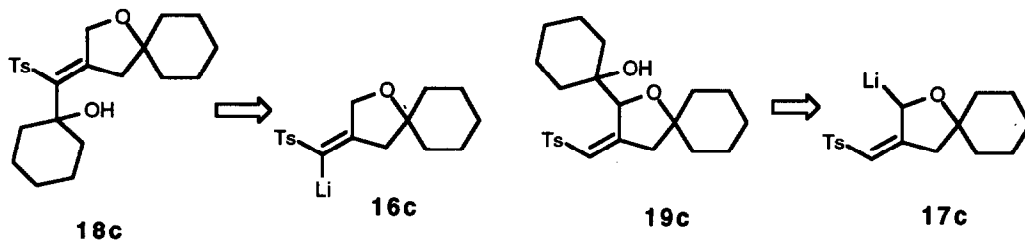
Scheme 5

Table 2. Reaction of Dianion **2** with Carbonyl Compounds. Synthesis of Furan Derivatives **6** and **7**.

Carbonyl compound ^a	Product				
	No.	R ¹	R ²	Yield (%) ^b	mp. ($^{\circ}\text{C}$) ^c or R_f ^d
MeCHO	6a	Me	H	62	85-86
EtCHO	6b	Et	H	68	0.47
Pr ⁱ CHO	6c	Pr ⁱ	H	59	0.48
Bu ^t CHO	15	-	-	55	72-73
PhCHO	6e	Ph	H	67	109-110 ^e
Et_2CO	7a	Et	Et	52	0.35
$(c\text{-C}_3\text{H}_5)_2\text{CO}$	7b	$c\text{-C}_3\text{H}_5$	$c\text{-C}_3\text{H}_5$	75	0.40
$(\text{CH}_2)_5\text{-C=O}$	7c ^f	$(\text{CH}_2)_5\text{-C}$		40 ^g	0.74 ^h
PhCOMe	7d	Ph	Me	61	107-109
Ph ₂ CO	7e	Ph	Ph	84	148-150

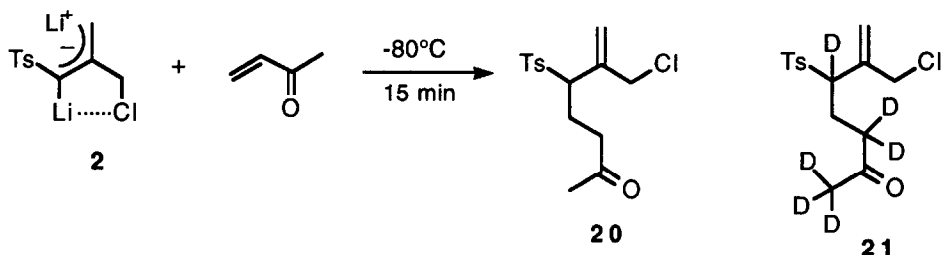
^a 1.2 Equivalents were added. ^b Isolated yield based on compound **1** after flash chromatography (silica gel). ^c Hexane/ether. ^d Hexane/ether: 1/1. ^e Lit.¹⁰ 110-112 $^{\circ}\text{C}$ (hexane/dichloromethane). ^f Compounds **18c** and **19c** were also obtained, the molar ratio being ca. 1:3:3 (^1H RMN). ^g Overall yield for compounds **7c**, **18c**, and **19c**. ^h R_f for compounds **7c**, **18c**, and **19c**, hexane/ether: 1/3.

The reaction of dianion **2** with cyclohexanone afforded a mixture of compounds **7c**, **18c** and **19c** in ca. 1: 3: 3 molar ratio. The formation of diaddition products **18c** and **19c** is due to the reaction of cyclohexanone with carbanions **16c** and **17c**, respectively (Scheme 6 and Table 2).



Scheme 6

When electrophilic olefins reacted with intermediate **2**, a Michael-induced ring closure¹⁵ process took place, and the corresponding tosylated cyclopentenes **8** were obtained (Scheme 1 and Table 3) after quenching at room temperature or at 0°C (see Table 3). In these cases dianion **2** reacted regioselectively at the α -position giving, with β -substituted olefins, a *cis/trans*-mixture of diastereomeric cyclopentenes **8b,c** (deduced by ¹H NMR) or only the *trans* one for compound **8d**, derived from cinnamionitrile. The configuration of compounds **8** was assigned according to nOe experiments. In the case of the reaction of dianion **2** with methyl vinyl ketone quenching was also carried out with water after 15 min at -80°C and the Michael adduct **20** was isolated in 85% yield (Scheme 7). When the quenching was carried out with deuterium oxide the hexadeuterated derivative **21** was obtained in 76% yield. The same Michael-type addition occurs in the reaction of monolithium derivative of the benzenesulfonyl homologue **1** with cycloalkenones but not ring closure was observed¹⁶. Following by TLC the reaction it can be observed that the cyclization process yielding compounds **8** starts at -20°C.



Scheme 7

The different behaviour of dianion **2** in cyclization process, depending on the type of electrophiles, is due to the stability of the corresponding cyclic compound. In the case of the reaction with alkyl halides to give cyclopropanes **5**, no isomerization of the double bond took place because methylenecyclopropanes are more stable than the isomeric methylcyclopropenes. In the reaction with aldehydes or electrophilic olefins the most stable dihydrofurans **6** or cyclopentenes **8** were, respectively, obtained and the isomerization of the *exo* double bond to the *endo* position favours also the cyclization process.

Table 3. Reaction of Dianion **2** with Electrophilic Olefins. Synthesis of Cyclopentenes **8**.

Electrophilic Olefin ^a	Product				
	No.	R	Z	Yield (%) ^b	mp. (°C) ^c or <i>R_f</i> ^d
CH ₂ =CHCOMe	8a	H	COMe	76	0.51 ^e
(<i>E</i>)-MeCH=CHCOMe ^f	8b	Me	COMe	78	0.24 ^g
(<i>E</i>)-MeCH=CHCO ₂ Me ^h	8c	Me	CO ₂ Me	62	0.26 ⁱ
(<i>E</i>)-PhCH=CHCN ^h	8d	Ph	CN	51	143-144 ^j

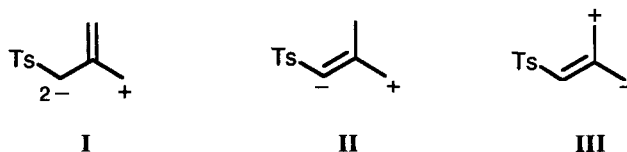
^a 1.2 Equivalents were added. ^b Isolated yield based on compound **1** after column chromatography on silica gel.

^c Hexane/ether. ^d Hexane/ether : 1/1. ^e Ether. ^f The reaction was quenched after 1 h at room temperature.

^g *Cis/trans* : 1/3. ^h The reaction was quenched at 0°C. ⁱ *Cis/trans* : 1/4. ^j Only *trans*.

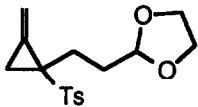
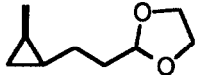
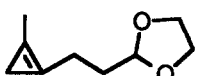
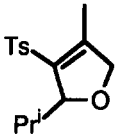
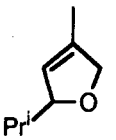
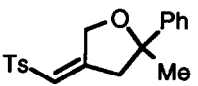
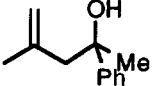
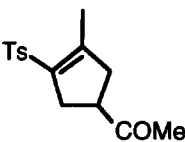
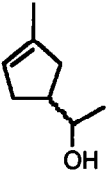
Reductive desulfonylation of representative compounds **5-8** was carried out with *ca.* 6% sodium amalgam¹⁷ in methanol in the presence of 4 equiv. of Na₂HPO₄ (Table 4). Tosylated methylenecyclopropane **5f** afforded a *ca.* 1:1 mixture of methylenecyclopropane **22**¹⁹ and cyclopropene **23** after 1 h at 0°C. The dihydrofuran **24** which was obtained by hydrogenolysis of compound **6c** after 2 h at 0°C, decomposed by chromatography or distillation. The reduction of methylenetetrahydrofurans **7d** was carried out with freshly prepared samarium(II) iodide²⁰ in THF-HMPA at -20°C during 1.5 h, affording the homoallylic alcohol **25** in 72% yield. The reduction of tetrahydrofurans **7** with sodium amalgam or sodium dithionite²¹ gave complex mixture of products or no reduction, respectively. Finally, the tosylated cyclopentene **8a** was reduced with sodium amalgam at room temperature for 1d to give a 1:1 mixture of diastereomeric cyclopentenylethanols **26**.

We can conclude that this chlorinated allyl sulfone dianion reacts regioselectively with electrophiles being specially useful in cycloannulation reactions acting as: (a) synthon **I** with alkyl halides, (b) synthon **II** with aldehydes and electrophilic olefins, and (c) synthon **III** with ketones (Scheme 8).

**Scheme 8****EXPERIMENTAL**

General. Melting points were obtained with a Reichert Thermovar apparatus and are uncorrected. IR spectra were obtained as films in a Pye Unicam SP3-200 spectrophotometer as neat liquids. ¹H and ¹³C spectra were recorded on a Bruker AC-300 spectrometer with SiMe₄ as internal standard and using CDCl₃ as solvent. ¹³C-NMR assignments were made on the basis of DEPT experiments. MS spectra were measured in a Hewlett-

Table 4. Reduction of Compounds 5-8 with Sodium Amalgam.

Starting Compound	Reaction conditions			Structure	Product		
	No.	T (°C)	Time		No.	Yield(%) ^a	<i>R_f</i> ^b
	5f	0	1 h		22	94 ^c	0.75
				23	0.75		
	6c	0	2 h		24	92 ^d	-e,f
	7ds	-20	1.5 h		25	72	0.40
	8a	25	1 d		26 ^h	90 ^d	0.30 ⁱ

^a Based on starting compound. ^b Hexane/ether:4/1. ^c Compounds **22** and **23** were obtained in 42:58 molar ratio (GLC) and could not be separated. ^d Crude pure compound. ^e *T_r*, 4.05 (see Experimental). ^f Lit.¹⁸. ^g The reduction was carried out with SnI_2 in THF-HMPA. ^h Mixture of diastereomers 1:1 deduced by GLC. ⁱ Hexane/ether:1/1.

Packard 5988A (EI, 70eV). High resolution mass spectra were measured in the corresponding Service at the University of Zaragoza. Elemental analyses were performed by the Microanalyses Service of the University of Alicante. Chromatographic analysis (GLC) were determined with a Hewlett-Packard HP-5890 instrument equipped with a 25 m WCOT capillary column (0.22 mm diam., 0.2 μm film thickness OV-101 stationary phase) using nitrogen (2 ml/min) as the carrier gas, $T_{\text{injector}}=270^\circ\text{C}$, $T_{\text{column}}=60^\circ\text{C}$, and 60-270 (15°C/min). Thin layer chromatography (TLC) was carried out on Schleicher & Schuell F1500/LS 254 plates coated with a 0.2 mm layer of silica gel and UV visualization. Column chromatography was performed using silica gel 60 of 70-270 mesh and hexane/ether as eluant. All starting materials were commercially available (Aldrich, Fluka, Janssen) of the best grade and were used without further purification. THF was dried with LiAlH_4 under argon atmosphere. Methanol was dried with magnesium turnings and stored under argon.

Reaction of Dianion 2 with Electrophiles. General Procedure.

To a solution of 2-(chloromethyl)-3-tosylpropene (1) (98 mg, 0.4 mmol) and *N,N*-dimethylpropyleneurea (106 μ l, 0.88 mmol) in THF (3 ml) cooled at -90°C was added a 1.6M solution of *n*-butyllithium (550 μ l, 0.88 mmol) in hexane. After 15 min stirring at the same temperature the electrophile was added (1.2 or 2.2 mmol, see tables 1-3) and the reaction was warmed up to room temperature (see tables 1-3)²². The reaction was hydrolyzed with 3M hydrochloric acid (4 ml) and extracted with ether (2x10 ml). The organic layer was dried over anhydrous Na_2SO_4 and evaporated *in vacuo* (15 torr). The crude product was purified by column chromatography and by recrystallization to afford compounds 3-9, 12, 15 and 18-21. Yields and physical data are included in tables 1-3, spectral and analytical data follow:

2-(Chloromethyl)-3,3-dideuterio-3-tosyl-1-propene (3): mp $85\text{--}86^{\circ}\text{C}$ (hexane/ether); ν (Nujol) 1640, (CH=C), 1300 and 1150 cm^{-1} (SO_2); δ_{H} 2.44 (s, 3H, CH_3Ar), 4.19 (s, 2H, CH_2Cl), 5.01, 5.44 (2s, 2H, $\text{CH}_2=\text{C}$), 7.35 and 7.74 (2d, $J=8.0$ Hz, 4H, ArH); δ_{C} 21.47 (CH_3Ar), 46.82 (CH_2Cl), 58.72 (quint., $J=21.8$ Hz, CD_2), 124.00, 133.38 ($\text{CH}_2=\text{C}$), 128.25, 129.66, 134.95 and 144.91 (ArC); m/z 246 (M^+ , 0.3%), 211 (25), 155 (44), 147 (12), 91 (100), 65 (17) and 55 (10). Anal. calcd. for $\text{C}_{11}\text{H}_{11}\text{ClD}_2\text{O}_2\text{S}$: C, 53.54; H, 4.49; and S, 12.99. Found: C, 53.51; H, 4.50; and S, 13.54.

2-(Chloromethyl)-3-methyl-3-tosyl-1-butene (4a): ν 3090, 1620, (HC=C), 1290 and 1140 cm^{-1} (SO_2); δ_{H} 1.51 [s, 6H, (CH_3)₂CS], 2.37 (s, 3H, CH_3Ar), 4.25 (s, 2H, CH_2Cl), 5.15, 5.58 (2s, 2H, $\text{CH}_2=\text{C}$), 7.26 and 7.59 (2d, $J=8.2$ Hz, 4H, ArH); δ_{C} 21.61 (CH_3Ar), 21.81 [(CH_3)₂CS], 44.84 (CH_2Cl), 66.19 (CS), 124.00, 141.95 ($\text{CH}_2=\text{C}$), 129.12, 130.71, 131.27 and 144.86 (ArC); m/z 236 ($M^+-\text{Cl}$, 2%), 157 (14), 119 (27), 117 (89), 92 (16), 91 (25), 81 (100), 79 (38), 67 (12), 65 (26), 53 (16) and 41 (16). Anal. calcd. for $\text{C}_{13}\text{H}_{17}\text{ClO}_2\text{S}$: C, 53.57; H, 6.28; and S, 11.75. Found: C, 53.58; H, 6.98; and S, 11.40.

3-Allyl-2-(chloromethyl)-3-tosyl-1,5-hexadiene (4b): ν 1300 and 1140 cm^{-1} (SO_2); δ_{H} 2.39 (s, 3H, CH_3Ar), 2.69 (dd, $J=15.2$ and 7.2 Hz, 2H, CH_2CS), 2.87 (dd, $J=15.2$ and 6.5 Hz, 2H, CH_2CS), 4.28 (s, 2H, CH_2Cl), 5.00-5.15 (m, 5H, $2\times\text{CH}_2=\text{CH}$ and 1H of $\text{CH}_2=\text{C}$), 5.68 (s, 1H, $\text{CH}_2=\text{C}$), 5.80 (m, 2H, $2\times\text{CH}=\text{CH}_2$), 7.25 and 7.60 (2d, $J=8.1$ Hz, 4H, ArH); δ_{C} 21.57 (CH_3Ar), 34.48 (CH_2CS), 44.28 (CH_2Cl), 71.56 (CS), 119.64, 124.60, 130.99, 131.43 ($3\times\text{C}=\text{C}$), 129.01, 130.72, 139.83 and 145.00 (ArC); m/z 169 ($M^+-\text{Ts}$, 5%), 133 (40), 119 (16), 105 (36), 93 (12), 92 (12), 91 (100), 79 (47), 77 (32), 65 (30) and 41 (24). Anal. calcd. for $\text{C}_{17}\text{H}_{21}\text{ClO}_2\text{S}$: C, 62.85; H, 6.52; and S, 9.87. Found: C, 63.02; H, 6.56; and S, 10.18.

5-(Chloromethyl)-4-propargyl-4-tosyl-5-hexen-1-yne (4c): *R_f* 0.62 (hexane/ether:1/1); ν (Nujol) 3290, 2120 ($\text{C}\equiv\text{CH}$), 3090, 1625 ($\text{CH}=\text{C}$), 1300 and 1140 cm^{-1} (SO_2); δ_{H} 2.05 (dt, $J=4.0$ and 2.5 Hz, 2H, $\text{HC}\equiv\text{C}$), 2.40 (s, 3H, CH_3Ar), 2.97, 3.15 (2dd, $J=17.3$ and 2.5 Hz, 4H, $\text{CH}_2\text{C}\equiv\text{C}$), 4.35 (s, 2H, CH_2Cl), 5.13, 5.80 (2s, 2H, $\text{CH}_2=\text{C}$), 7.29 and 7.66 (2d, $J=8.2$ Hz, 4H, ArH); δ_{C} 21.62 (CH_3Ar), 22.36 ($\text{CH}_2\text{C}\equiv\text{C}$), 44.12 (CH_2Cl), 69.85 (CS), 72.80, 77.31 ($\text{C}\equiv\text{C}$), 124.30, 130.94 ($\text{CH}_2=\text{C}$), 129.33, 130.78, 138.13 and 145.69 (ArC); m/z 321 (M^+ , 0.2%), 157 (35), 139 (29), 130 (13), 129 (55), 128 (100), 127 (38), 116 (14), 115 (52), 92 (14), 91 (49), 89 (14), 77 (19), 65 (31), 63 (15) and 51 (11). Anal. calcd. for $\text{C}_{17}\text{H}_{17}\text{ClO}_2\text{S}$: C, 63.63; H, 5.34; and S, 9.99. Found: C, 63.65; H, 5.26; and S, 9.90.

3-Benzyl-2-(chloromethyl)-4-phenyl-3-tosyl-1-butene (4d): ν 3085, 1620 (CH=C), 1300 and 1140 cm^{-1} (SO_2); δ_{H} 2.39 (s, 3H, CH_3Ar), 3.10, 3.72 (2d, $J=16.0$ Hz, 4H, $2\times\text{CH}_2\text{Ph}$), 4.31 (s, 2H, CH_2Cl), 5.14, 5.75 (2s, 2H, $\text{CH}_2=\text{C}$), 7.05-7.39 (m with d at 7.24, $J=8.1$ Hz, 12 H, $2\times\text{Ph}$ and 2H *p*-Tol), and 7.65 (d, $J=8.1$ Hz, 2H, *p*-TolH); δ_{C} 21.62 (CH_3Ar), 34.87 (CH_2Ph), 45.02 (CH_2Cl), 74.44 (CS), 125.44, 131.57 ($\text{CH}_2=\text{C}$), 126.64, 127.91, 129.05, 130.43, 131.03, 135.18, 140.11 and 144.98 (ArC); m/z 389 (M^+ -35, 1%), 233 (10), 232 (10), 165 (10), 91 (100), 77 (10) and 65 (10).

3-Chloro-2-(1-tosylcyclopentyl)-1-propene (4e): ν 3075, 1640 (CH=C), 1300 and 1140 cm^{-1} (SO_2); δ_{H} 1.45-2.20 [m, 6H, (CH_2)₃CS], 2.13-2.48 (m with s at 2.43, 5H, CH_2CS and CH_3Ar), 4.19 (s, 2H, CH_2Cl), 5.07, 5.62 (2s, 2H, $\text{CH}_2=\text{C}$), 7.28 and 7.64 (2d, $J=8.0$ Hz, 4H, ArH); δ_{C} 21.62 (CH_3Ar), 24.38, 32.54 ($\text{CH}_2\text{CH}_2\text{CS}$), 45.26 (CH_2Cl), 77.61 (CS), 124.55, 132.31 ($\text{CH}_2=\text{C}$), 129.07, 130.49, 141.76 and 144.68 (ArC); m/z 198 (M^+ - $\text{SO}_2\text{-Cl}$, 7%), 157 (19), 145 (17), 144 (13), 143 (67), 139 (15), 107 (100), 91 (92), 79 (93), 77 (40), 65 (58) and 47 (27).

1-Ethyl-1-tosyl-2-methylenecyclopropane (5a): ν 1300 and 1140 cm^{-1} (SO_2); δ_{H} 0.78 (t, $J=7.5$ Hz, 3H, CH_3CH_2), 1.45 (dt, $J=10.0$ and 2.1 Hz, 1H, $\text{CH}_2\text{C}=\text{C}$), 1.82 (m, 2H, CH_2CH_3), 2.21 (m, 1H, $\text{CH}_2\text{C}=\text{C}$), 2.39 (s, 3H, CH_3Ar), 5.52, 5.60 (2t, $J=2.3$ Hz, 2H, $\text{CH}_2=\text{C}$), 7.30 and 7.71 (2d, $J=8.2$ Hz, 4H, ArH); δ_{C} 9.87 (CH_3CH_2), 13.26 ($\text{CH}_2\text{C}=\text{C}$), 21.48 (CH_3Ar), 22.87 (CH_2CH_3), 44.74 (CS), 107.36, 129.44 ($\text{CH}_2=\text{C}$), 128.77, 129.50, 135.39 and 144.32 (ArC); m/z 236 (M^+ , 13%), 157 (27), 143 (18), 140 (15), 139 (100), 105 (21), 92 (24), 91 (51), 81 (24), 79 (53), 77 (20), 65 (36) and 53 (28). High resolution mass spectrum required for $\text{C}_{13}\text{H}_{16}\text{O}_2\text{S}$: m/z 236.0871. Found: 236.0876.

1-Butyl-1-tosyl-2-methylenecyclopropane (5b): ν 1300 and 1150 cm^{-1} (SO_2); δ_{H} 0.76 (deform. t, 3H, CH_3CH_2), 1.15 (m, 4H, $\text{CH}_2\text{CH}_2\text{CH}_3$), 1.45, 2.21 (2dt, $J=10.0$ and 2.5 Hz, 2H, $\text{CH}_2\text{C}=\text{C}$), 1.75 (m, 2H, $\text{CH}_2\text{CH}_2\text{CS}$), 2.40 (s, 3H, CH_3Ar), 5.52, 5.60 (2t, $J=2.5$ Hz, 2H, $\text{CH}_2=\text{C}$), 7.30 and 7.71 (2d, $J=8.2$ Hz, 4H, ArH); δ_{C} 13.59 (CH_3CH_2), 13.94 ($\text{CH}_2\text{C}=\text{C}$), 21.45 (CH_3Ar), 22.36 (CH_2CH_3), 27.68 ($\text{CH}_2\text{CH}_2\text{CH}_3$), 29.52 ($\text{CH}_2\text{CH}_2\text{CS}$), 44.00 (CS), 107.27, 129.77 ($\text{CH}_2=\text{C}$), 128.74, 129.46, 135.34 and 144.28 (ArC); m/z 264 (M^+ , 0.3%), 157 (19), 143 (30), 142 (11), 140 (13), 139 (100), 109 (30), 93 (21), 92 (21), 91 (51), 79 (27), 77 (20), 67 (55), 65 (39), 55 (20) and 41 (30). Anal. calcd. for $\text{C}_{15}\text{H}_{20}\text{O}_2\text{S}$: C, 68.14; H, 7.62; and S, 12.13. Found: C, 68.37; H, 8.07; and S, 12.56.

1-(Trimethylsilylmethyl)-1-tosyl-2-methylenecyclopropane (5c): ν 3075, 1640, (CH=C), 1300, 1150 cm^{-1} (SO_2); δ_{H} -0.05 [s, 9H, (CH_3)₃Si], 1.28 (d, $J=7.6$ Hz, 2H, CH_2Si), 1.50, 2.28 (2d, $J=10.0$ Hz, 2H, $\text{CH}_2\text{C}=\text{C}$), 2.43 (s, 3H, CH_3Ar), 5.60, 5.65 (2s, 2H, $\text{CH}_2=\text{C}$), 7.33 and 7.79 (2d, $J=7.9$ Hz, 4H, ArH); δ_{C} -0.52 [(CH_3)₃Si], 15.05 ($\text{CH}_2\text{C}=\text{C}$), 15.26 (CH_2Si), 21.54 (CH_3Ar), 42.34 (CS), 107.91 ($\text{CH}_2=\text{C}$), 129.08, 129.44, 131.22, 134.78 and 144.19 (ArH and $\text{C}=\text{CH}_2$); m/z 293 (M^+ , 0.7%), 279 (2), 157 (37), 149 (37), 142 (44), 139 (33), 91 (19), 75 (12), 73 (100), 65 (18) and 41 (16).

1-(2-Tetrahydropyranoxyethyl)-1-tosyl-2-methylenecyclopropane (5d): ν 3085, 1640 (CH=C), 1300 and 1150 cm^{-1} (SO_2); δ_{H} 1.40-2.10 (m, 9H, $3\times\text{CH}_2\text{CO}$, CH_2CCO and 1H of $\text{CH}_2\text{C}=\text{C}$), 2.13 (d, $J=7.0$ Hz, 1H, $\text{CH}_2\text{C}=\text{C}$), 2.34 (s, 3H, CH_3Ar), 3.35-3.65 (m, 4H, $2\times\text{CH}_2\text{O}$), 4.40 (deform. d, $J=13.5$ Hz, 1H, CHO), 5.59

(d, $J=2.2$ Hz, 1H, $\text{CH}_2=\text{C}$), 5.67 (s, 1H, $\text{CH}_2=\text{C}$), 7.34 and 7.77 (2d, $J=8.0$ Hz, 4H, ArH); δ_{C} 14.80 ($\text{CH}_2\text{C}=\text{C}$), 19.41 (CH_2CCO), 21.57 (CH_3Ar), 25.29, 29.95, 30.49 ($3\times\text{CH}_2\text{CCO}$), 42.23 (CS), 62.23, 64.34 ($2\times\text{CH}_2\text{O}$), 98.85 (CHO), 107.75, 129.47 ($\text{CH}_2=\text{C}$), 128.91, 129.62, 135.25 and 144.47 (ArC); m/z 253 ($M^+ + 2\text{-THP}$, 0.5%), 251 ($M^+ - \text{THP}$, 0.6), 2.35 (12), 157 (11), 149 (14), 139 (43), 92 (15), 91 (65), 85 (100), 79 (24), 78 (10), 77 (39), 67 (29), 65 (58), 63 (16), 57 (30), 56 (18), 55 (35), 53 (20), 52 (17), 43 (29) and 41 (85).

1-(2-Hydroxyethyl)-1-tosyl-2-methylenecyclopropane (5e): ν 3450 (OH), 1300 and 1150 cm^{-1} (SO_2); δ_{H} 1.65 (dt, $J=10.0$ and 2.1 Hz, 1H, $\text{CH}_2\text{C}=\text{C}$), 2.00 (m, 2H, $\text{CH}_2\text{CH}_2\text{O}$), 2.31 (dt, $J=10.0$ and 2.3 Hz, 1H, $\text{CH}_2\text{C}=\text{C}$), 2.39 (s, 3H, CH_3Ar), 2.75 (br. s, 1H, OH), 3.72 (m, 2H, CH_2O), 5.63 (m, 2H, $\text{CH}_2=\text{C}$), 7.37 and 7.76 (2d, $J=8.0$ Hz, 4H, ArH); δ_{C} 15.55 ($\text{CH}_2\text{C}=\text{C}$), 21.55 (CH_3Ar), 33.57 ($\text{CH}_2\text{CH}_2\text{O}$), 42.64 (CS), 59.99 (CH_2O), 107.90, 129.55 ($\text{CH}_2=\text{C}$), 128.89, 129.70, 134.59 and 144.79 (ArC); m/z 251 ($M^+ - 1$, 0.5%), 221 (18), 157 (26), 149 (90), 143 (38), 142 (29), 141 (18), 141 (21), 139 (100), 129 (10), 121 (16), 105 (23), 97 (60), 92 (51), 91 (91), 89 (25), 85 (31), 79 (16), 77 (56), 67 (24), 65 (78), 63 (26), 55 (14), 53 (18), 51 (23), 43 (23) and 41 (37).

2-[2-(2-Methylene-1-tosylcyclopropyl)ethyl]-1,3-dioxolane (5f): ν 1300 and 1145 cm^{-1} (SO_2); δ_{H} 1.47 (d, $J=10.0$ Hz, 1H, $\text{CH}_2\text{C}=\text{C}$), 1.64 (m, 2H, $\text{CH}_2\text{CH}_2\text{CO}$), 1.86 (m, 2H, CH_2CO), 2.25 (d, $J=10.0$ Hz, 1H, $\text{CH}_2\text{C}=\text{C}$), 2.41 (s, 3H, CH_3Ar), 3.76 (m, 4H, $2\times\text{CH}_2\text{O}$), 4.73 (t, $J=4.2$ Hz, 1H, CHO), 5.55, 5.60 (2s, 2H, $\text{CH}_2=\text{C}$) 7.30 and 7.70 (2d, $J=7.9$ Hz, 4H, ArH); δ_{C} 14.31 ($\text{CH}_2\text{C}=\text{C}$), 21.50 (CH_3Ar), 24.24 ($\text{CH}_2\text{CH}_2\text{CO}$), 30.16 ($\text{CH}_2\text{CH}_2\text{CO}$), 43.43 (CS), 64.67 (CH_2O), 103.24 (CHO), 107.54, 129.61 ($\text{CH}_2=\text{C}$), 128.79, 129.57, 135.19 and 144.42 (ArC); m/z 308 (M^+ , 0.2%), 153 (12), 139 (18), 91 (25), 86 (12), 81 (13), 79 (15), 77 (11), 73 (100), 65 (18) and 45 (22). High resolution mass spectrum required for $\text{C}_{16}\text{H}_{20}\text{O}_4\text{S}$: m/z 308.1083. Found: 308.1036.

1-Benzyl-1-tosyl-2-methylenecyclopropane (5g): ν 1300 and 1145 cm^{-1} (SO_2); δ_{H} 1.13, 2.16 (2dd, $J=10.0$ and 2.1 Hz, 2H, $\text{CH}_2\text{C}=\text{C}$), 2.39 (s, 3H, CH_3Ar), 3.17 (s, 2H, CH_2Ph), 5.51 (m, 2H, $\text{CH}_2=\text{C}$), 6.92 (m, 2H, Ph), 7.12 (m, 3H, Ph), 7.27 and 7.69 (2d, $J=7.9$ Hz, 4H, *p*-Tol); δ_{C} 13.13 ($\text{CH}_2\text{C}=\text{C}$), 21.53 (CH_3Ar), 34.98 (CH_2Ph), 44.36 (CS), 108.38, 134.83 ($\text{CH}_2=\text{C}$), 126.82, 127.91, 128.52, 128.89, 129.50, 129.90, 135.21 and 144.37 (ArC); m/z 298 (M^+ , 0.5%), 143 (67), 142 (65), 141 (71), 139 (24), 129 (15), 128 (76), 127 (14), 115 (34), 92 (12), 91 (100), 65 (37) and 62 (10).

2,4-Dimethyl-3-tosyl-2,5-dihydrofuran (6a): R_f 0.32 (hexane/ether:1/1); ν 3070, 1640 ($\text{CH}=\text{C}$), 1300 and 1150 cm^{-1} (SO_2); δ_{H} 1.36 (d, $J=6.4$ Hz, 3H, CH_3CO), 2.12 (s, 3H, CH_3CCH_2), 2.41 (s, 3H, CH_3Ar), 4.55, 4.64 (2dd, $J=14.2$ and 6.9 Hz, 2H, CH_2O), 4.98 (m, 1H, CHO), 7.35 and 7.77 (2d, $J=8.2$ Hz, 4H, ArH); δ_{C} 11.16 (CH_3CCH_2), 21.34, 21.50 (CH_3CO and CH_3CO), 78.17 (CH_2O), 82.96 (CHO), 128.31, 129.81, 135.45, 138.46, 144.47 and 149.69 (ArC and $\text{C}=\text{C}$); m/z 237 ($M^+ - 15$, 11), 97 (17), 92 (15), 91 (100), 89 (66), 67 (20), 65 (72), 63 (31), 53 (70), 51 (23), 43 (39) and 41 (31).

2-Ethyl-4-methyl-3-tosyl-2,5-dihydrofuran (6b): ν 3070, 1640 ($\text{CH}=\text{C}$), 1300 and 1150 cm^{-1} (SO_2); δ_{H} 0.82 (t, $J=7.3$ Hz, 3H, CH_3CH_2), 1.66, 1.85 (2m, 2H, CH_2CH_2), 2.13 (s, 3H, $\text{CH}_3\text{C}=\text{CH}_2$), 2.40 (s, 3H, CH_3Ar), 4.56 (dd, $J=14.5$ and 2.0 Hz, 1H, CH_2O), 4.62 (dd, $J=14.5$ and 5.5 Hz, 1H, CH_2O), 4.90 (m, 1H, CHO), 7.33

and 7.76 (2d, $J=8.0$ Hz, 4H, ArH); δ_c 8.43 (CH_3CH_2), 11.17 (CH_3CCH_2), 21.49 (CH_3Ar), 27.65 (CH_2CH_3), 78.97 (CH_2O), 87.64 (CHO), 127.15, 129.79, 133.56, 138.32, 144.46 and 150.38 (ArC and C=C); m/z 238 ($M^+ + 1\text{-Et}$, 11%), 237 ($M^+ - \text{Et}$, 73), 236 (13), 139 (17), 92 (17), 91 (100), 79 (11), 77 (14), 69 (13), 67 (11), 65 (37), 53 (20) and 43 (47). Anal. calcd. for $\text{C}_{14}\text{H}_{18}\text{O}_3\text{S}$: C, 63.13; H, 6.81; and S, 12.04. Found: C, 63.07; H, 6.79; and S, 11.99.

2-Ethyl-4-(deuteriomethyl)-3-tosyl-2,5-dihydrofuran (6b'): R_f 0.47 (hexane/ether:1/1); ν 3070, 1640, (CH=C), 1300 and 1150 cm^{-1} (SO_2); δ_H 0.82 (t, $J=7.3$ Hz, 3H, CH_3CH_2), 1.66, 1.85 (2m, 2H, CH_2CH_3), 2.13 (m, 2H, CH_2D), 2.40 (s, 3H, CH_3Ar), 4.56 (dd, $J=14.5$ and 2.0 Hz, 1H, CH_2O), 4.62 (dd, $J=14.5$ and 5.5 Hz, 1H, CH_2O), 4.90 (m, 1H, CHO), 7.33 and 7.76 (2d, $J=8.0$ Hz, 4H, ArH); δ_c 8.43 (CH_3CH_2), 10.99 (t, $J=19.5$ Hz, CD), 21.49 (CH_3Ar), 27.65 (CH_2CH_3), 78.97 (CH_2O), 87.64 (CHO), 127.15, 129.79, 133.56, 138.82, 144.46 and 150.38 (ArC and C=C); m/z 238 ($M^+ - \text{Et}$, 5%), 237 (41), 155 (19), 139 (11), 92 (13), 91 (100), 89 (17), 77 (11), 65 (42), 53 (32), 51 (14) and 41 (21).

2-Isopropyl-4-methyl-3-tosyl-2,5-dihydrofuran (6c): ν 3070, 1640 (CH=C), 1300 and 1150 cm^{-1} (SO_2); δ_H 0.64, 0.98 [2d, $J=7.0$ Hz, 6H, $(\text{CH}_3)_2\text{CH}$], 2.10-2.20 (m with s at 2.14, 4H, CHCH= and CH_3CCH_2), 2.39 (s, 3H, CH_3Ar), 4.60 (deform. d, $J=14.5$ Hz, 2H, CH_2O), 4.83 (m, 1H, CHO), 7.34 and 7.77 (2d, $J=8.1$ Hz, 4H, ArH); δ_c 11.22 (CH_3CCH_2), 14.16, 19.85 [$(\text{CH}_3)_2\text{CH}$], 21.57 (CH_3Ar), 31.99 (CHCHO), 79.96 (CH_2O), 127.31, 129.83, 133.15, 138.27, 144.52 and 150.50 (ArC and C=C); m/z 237 ($M^+ - \text{Pr}^i$, 38%), 128 (13), 92 (12), 91 (98), 77 (16), 70 (13), 67 (13), 65 (53), 55 (26), 43 (99) and 41 (100).

(Z)-2,2-Diethyl-4-(tosylmethylene)tetrahydrofuran (7a): ν 3070, 1620 (CH=C), 1300 and 1150 cm^{-1} (SO_2); δ_H 0.85 (t, $J=7.5$ Hz, 6H, $2\times\text{CH}_3\text{CH}_2$), 1.39-1.96 (m, 4H, $2\times\text{CH}_2\text{CH}_3$), 2.40 (s, 3H, CH_3Ar), 2.52 (s, 2H, CH_2CO), 4.44 (s, 2H, CH_2O), 6.05 (s, 1H, CH=C), 7.32 and 7.77 (2d, $J=8.1$ Hz, 4H, ArH); δ_c 7.72 (CH_3CH_2), 21.56 (CH_3Ar), 30.70 (CH_2CH_3), 42.15 (CH_2CO), 57.30 (CH_2O), 75.20 (CO), 122.14, 128.81 (CH=C), 128.28, 129.59, 135.91 and 144.73 (ArC); m/z 209 ($M^+ - \text{Et}_2\text{CO}$, 34%), 155 (16), 145 (21), 139 (21), 109 (15), 91 (100), 89 (74), 87 (47), 69 (17), 65 (57), 57 (70), 53 (37), 45 (61), 43 (22) and 41 (33).

(Z)-5-Deuterio-2,2-diethyl-4-(tosylmethylene)tetrahydrofuran (7a''): R_f 0.35 (hexane/ether:1/1); ν 3070, 1620 (CH=C), 1300 and 1150 cm^{-1} (SO_2); δ_H 0.85 (t, $J=7.5$ Hz, 6H, $2\times\text{CH}_3\text{CH}_2$), 1.39-1.96 (m, 4H, $2\times\text{CH}_2\text{CH}_3$), 2.40 (s, 3H, CH_3Ar), 2.52 (s, 2H, CH_2CO), 4.44 (s, 1H, CHD), 6.05 (s, 1H, CH=C), 7.32 and 7.77 (2d, $J=8.1$ Hz, 4H, ArH); δ_c 7.72 (CH_3CH_2), 21.56 (CH_3Ar), 30.70 (CH_2CH_3), 42.15 (CH_2CO), 57.04 (t, $J=21.0$ Hz, CHD), 75.20 (CO), 122.14, 128.81 (CH=C), 128.28, 129.59, 135.91 and 144.73 (ArC); m/z 210 ($M^+ - \text{Et}_2\text{CO}$, 7%), 209 (6), 155 (17), 139 (15), 92 (21), 91 (100), 87 (37), 77 (10), 69 (13), 65 (52), 57 (68), 55 (12), 53 (14), 45 (48) and 41 (23).

(Z)-2,2-Dicyclopropyl-4-(tosylmethylene)tetrahydrofuran (7b): ν 3080, 1625 (CH=C), 1300 and 1150 cm^{-1} (SO_2); δ_H 0.29-0.41 (m, 8H, $4\times\text{CH}_2\text{CH}$), 0.71-0.79 (m, 2H, $2\times\text{CHCH}_2$), 2.40 (s, 3H, CH_3Ar), 2.59 (s, 2H, CH_2CO), 4.38 (s, 2H, CH_2O), 6.09 (s, 1H, CH=C), 7.32 and 7.76 (2d, $J=8.1$ Hz, 4H, ArH); δ_c -0.23, 0.73 (CH_2CH), 18.59 (CHCH₂), 21.52 (CH_3Ar), 45.07 (CH_2CO), 57.84 (CH_2O), 71.46 (CO), 122.34, 128.36, 129.52, 129.53, 135.75 and 144.67 (C=C and ArC); m/z 318 (M^+ , 0.1%), 111 (100), 91 (27), 89 (10), 69 (81), 65 (15),

63 (14) and 41 (42).

(Z)-2-Phenyl-2-methyl-4-(tosylmethylene)tetrahydrofuran (7d): ν (Nujol) 1300 and 1140 cm^{-1} (SO_2); δ_{H} 1.56 (s, 3H, CH_3CO), 2.40 (s, 3H, CH_3Ar), 2.78, 2.92 (2d, $J=14.5$ Hz, 2H, CH_2CO), 3.80, 4.13 (2d, $J=13.7$ Hz, 2H, CH_2O), 5.89 (s, 1H, $\text{CH}=\text{C}$), 7.20-7.39 (m, 7H, Ph and 2*xp*-Tol) and 7.67 (d, $J=8.0$ Hz, 2H, *p*-TolH); δ_{C} 21.52 (CH_3Ar), 30.30 (CH_3CO), 47.31 (CH_2CO), 57.06 (CH_2O), 74.49 (CO), 123.42, 124.64, 126.86, 127.69, 128.19, 128.29, 129.50, 135.59, 144.69 and 146.35 (ArC and $\text{C}=\text{C}$); m/z 209 (M^+ -PhCOMe, 4%), 155 (12), 121 (77), 105 (14), 91 (68), 89 (22), 77 (24), 65 (38) and 43 (100). Anal. calcd. for $\text{C}_{19}\text{H}_{20}\text{O}_3\text{S}$: C, 69.48; H, 6.14; and S, 9.76. Found: C, 69.60; H, 5.97; and S, 9.86.

(Z)-2,2-Diphenyl-4-(tosylmethylene)tetrahydrofuran (7e): ν (Nujol) 1620 ($\text{CH}=\text{C}$), 1300 and 1150 cm^{-1} (SO_2); δ_{H} 2.41 (s, 3H, CH_3Ar), 3.39 (s, 2H, CH_2CO), 3.90 (s, 2H, CH_2O), 5.82 (s, 1H, $\text{CH}=\text{C}$), 7.21-7.38 (m, 12H, 2*x*Ph and 2*xp*-Tol), 7.68 (d, $J=8.1$ Hz, 2H, 2*xp*-TolH); δ_{C} 21.56 (CH_3Ar), 45.13 (CH_2CO), 57.14 (CH_2O), 77.81 (CO), 124.11, 125.83, 126.01, 126.97, 127.17, 128.08, 128.11, 128.18, 128.32, 129.56, 129.73, 135.70, 144.73 and 145.83 (ArC and $\text{C}=\text{C}$); m/z 313 (M^+ -Ph, 4%), 235 (12), 183 (100), 157 (14), 105 (99), 91 (39), 89 (12), 77 (54), 65 (16) and 51 (11). Anal. calcd. for $\text{C}_{24}\text{H}_{22}\text{O}_3\text{S}$: C, 73.82; H, 5.68; and S, 8.21. Found: C, 74.00; H, 5.47; and S, 8.20.

4-Acetyl-1-methyl-2-tosylcyclopentene (8a): ν 3060, 1640 ($\text{CH}=\text{C}$), 1710 ($\text{C}=\text{O}$), 1300 and 1150 cm^{-1} (SO_2); δ_{H} 2.12 (s, 3H, CH_3CO), 2.16 (t, $J=1$ Hz, 3H, CH_3CCH_2), 2.40 (s, 3H, CH_3Ar), 2.68-2.92 (m, 4H, 2*x* CH_2CO), 3.17-3.24 (m, 1H, CHCO), 7.33 and 7.75 (2d, $J=8.1$ Hz, 4H, ArH); δ_{C} 14.99 (CH_3CCH_2), 21.51 (CH_3Ar), 28.24 (CH_3CO), 35.84, 41.92 (2*x* CH_2CO), 47.01 (CHCO), 127.19, 129.77, 132.29, 137.82, 144.18, 151.99 (ArC and $\text{C}=\text{C}$) and 207.38 (CO); m/z 278 (M^+ , 0.8%), 235 (15), 157 (12), 139 (18), 123 (18), 91 (26), 79 (39), 77 (24), 65 (18) and 43 (100). High resolution mass spectrum required for $\text{C}_{15}\text{H}_{18}\text{O}_3\text{S}$: m/z 278.0977. Found: 278.0965.

cis/trans-4-Acetyl-1,3-dimethyl-2-tosylcyclopentene (8b): ν 3050, 1630 ($\text{CH}=\text{C}$), 1705 (CO), 1300 and 1145 cm^{-1} (SO_2); δ_{H} (*cis/trans*) 0.87, 1.23 (2d, $J=6.5$ Hz, 3H, CH_3CH), 1.98 (s, 3H, CH_3CO), 2.05 (s, 3H, CH_3CCH_2), 2.20-2.45 (m with s at 2.35, 4H, CH_3Ar and CHCH_3), 2.65 (s, 2H, CH_2), 3.07 (m, 1H, CHCO), 7.23 and 7.65 (2d, $J=8.1$ Hz, 4H, ArH); δ_{C} (*trans*) 15.41 (CH_3CCH_2), 21.03 (CH_3CH), 21.42 (CH_3Ar), 27.91 (CH_3CO), 39.98 (CH_2), 44.51 (CHCH_3), 56.00 (CHCO), 126.87, 129.54, 137.10, 138.74, 143.84, 152.64 (ArC and $\text{C}=\text{C}$) and 207.03 (CO); δ_{C} (*cis*) 14.80, 15.63 (CH_3CCH_2 , CH_3CH), 21.43 (CH_3Ar), 29.45 (CH_3CO), 38.27 (CH_2), 42.60 (CHCH_3), 54.18 (CHCO), 126.98, 129.68, 137.96, 138.87, 143.92, 153.30 (ArC and $\text{C}=\text{C}$) and 206.28 (CO); m/z 292 (M^+ , 1%), 250 (28), 249 (19), 157 (13), 139 (27), 137 (25), 95 (14), 94 (12), 93 (60), 92 (11), 91 (44), 79 (23), 78 (11), 77 (35), 65 (23), 53 (11) and 43 (100).

cis/trans-1,3-Dimethyl-4-(methoxycarbonyl)-2-tosylcyclopentene (8c): ν 3080, 1630 ($\text{CH}=\text{C}$), 1730 (CO), 1300 and 1150 cm^{-1} (SO_2); δ_{H} 1.28 (d, $J=6.5$ Hz, 3H, CH_3CH), 2.15 (t, $J=0.8$ Hz, 3H, CH_3CCH_2), 2.43 (s, 3H, CH_3Ar), 2.61 (qd, $J=6.5$ and 4.7 Hz, 1H, CHCH_3), 2.79 (m, 2H, CH_2), 3.22 (m, 1H, CHCO), 3.60 (s, 3H, CH_3O), 7.31 and 7.74 (2d, $J=8.2$ Hz, 4H, ArH); δ_{C} 15.38 (CH_3CCH_2), 20.34 (CH_3CH), 21.36 (CH_3Ar), 41.24 (CH_2), 45.75 (CHCH_3), 48.19 (CHCO), 51.77 (CH_3O), 127.07, 129.57, 129.62, 137.35, 138.63, 143.87

(ArC and C=C) and 174.30 (CO); m/z 249 ($M^+ + 1$ -CO₂Me, 5%), 248 (M^+ -CO₂Me, 19), 152 (11), 139 (38), 93 (92), 92 (80), 91 (100), 77 (71), 65 (58), 55 (29) and 41 (46).

trans-4-Cyano-3-phenyl-1-methyl-2-tosylcyclopentene (8d): ν (CHCl₃) 3050, 1630 (CH=C), 2220 (C≡N), 1300 and 1140 cm⁻¹ (SO₂); δ_{H} 2.33 (t, $J=0.7$ Hz, 3H, CH₃CCH₂), 2.38 (s, 3H, CH₃Ar), 2.89 [ddq, $J_{\text{a,b}}=17.0$ Hz, $J_{\text{a,c}}=4.0$ Hz and $J_{\text{a,Me}}=0.7$ Hz, 1H, CH₂(a)], 2.96 (ddd, $J_{\text{c,d}}=4.0$ Hz, $J_{\text{c,b}}=8$ Hz and $J_{\text{c,d}}=8.0$ Hz, 1H, CHCN), 3.12 [ddq, $J_{\text{b,a}}=17.0$ Hz, $J_{\text{b,c}}=8.0$ Hz and $J_{\text{b,Me}}=0.7$ Hz, 1H, CH₂(b)], 4.46 (m, 1H, CHPh) and 6.93-7.75 (m, 9H, Ph and *p*-TolH); δ_{C} 15.58 (CH₃CCH₂), 21.44 (CH₃Ar), 34.72 (CHCN), 43.20 (CH₂), 57.94 (CHPh), 120.97 (CN), 127.34, 127.49, 127.79, 128.72, 129.33, 137.11, 137.75, 138.84, 144.03 and 153.37 (ArC and C=C); m/z 337 (M^+ , 0.1%), 209 (16), 155 (18), 153 (34), 152 (43), 139 (13), 121 (20), 94 (12), 93 (85), 92 (25), 91 (100), 89 (18), 79 (20), 77 (27), 63 (11), 53 (21) and 51 (10). Anal. calcd. for C₂₀H₁₉NO₂S: C, 71.19; H, 5.68; N, 4.15; and S, 9.50. Found: C, 71.17; H, 5.63; N, 4.05; and S, 9.41.

2-(Chloromethyl)-3-tosyl-1-butene (9a): ν 3085, 1630 (CH=C), 1300 and 1140 cm⁻¹ (SO₂); δ_{H} 1.43 (d, $J=7.2$ Hz, 3H, CH₃CS), 2.39 (s, 3H, CH₃Ar), 3.91 (q, $J=7.2$ Hz, 1H, CHS), 4.06, 4.18 (2d, $J=12$ Hz, 2H, CH₂Cl), 5.18, 5.48 (2s, 2H, CH₂=C), 7.30 and 7.67 (2d, $J=8.1$ Hz, 4H, ArH); δ_{C} 14.79 (CH₃CS), 21.59 (CH₃Ar), 48.29 (CH₂Cl), 61.88 (CHS), 121.59, 139.01 (CH₂=C), 129.32, 129.55, 133.51 and 144.93 (ArC); m/z 223 (M^+ -Cl, 4%), 157 (30), 156 (22), 139 (10), 177 (10), 103 (32), 102 (17), 92 (51), 91 (41), 67 (100), 65 (38) and 41 (22).

2-(Chloromethyl)-3-tosyl-1,5-hexadiene (9b): ν 3060, 1640 (CH=C), 1300 and 1140 cm⁻¹ (SO₂); δ_{H} 2.50-2.35 (m with s at 2.40, 4H, CH₃Ar and 1H CH₂CS), 3.95 (m, 1H, CH₂CS), 3.81 (m, 1H, CHS), 4.05, 4.06 (2d, $J=12.0$ Hz, 2H, CH₂Cl), 5.00-5.14 (m, 2H, CH₂=CH), 5.14, 5.54 (2s, 2H, CH₂=C), 5.56 (m, 1H, CH=CH₂), 7.30 and 7.68 (2d, $J=8.1$ Hz, 4H, ArH); δ_{C} 21.60 (CH₃Ar), 32.38 (CH₂CS), 48.14 (CH₂Cl), 66.54 (CS), 118.46, 122.26, 138.38, 133.57 (2x C=C), 129.38, 129.58, 136.32 and 145.05 (ArC); m/z 249 (M^+ -Cl, 3%), 157 (12), 130 (11), 129 (16), 128 (28), 93 (100), 92 (16), 91 (90), 79 (51), 78 (10), 77 (65), 65 (38), 53 (14), 51 (10) and 41 (16).

2-(Chloromethyl)-4-phenyl-3-tosyl-1-butene (9d): ν 3085, 1630, 800 (CH=C), 1300 and 1145 cm⁻¹ (SO₂); δ_{H} 2.40 (s, 3H, CH₃Ar), 2.94 (dd, $J=14.0$ and 11.4 Hz, 1H, CH₂Ph), 3.46 (dd, $J=14.0$ and 3.5 Hz, 1H, CH₂Ph), 3.72, 3.90 (2d, $J=12.4$ Hz, 2H, CH₂Cl), 4.00 (dd, $J=11.4$ and 3.5 Hz, 1H, CHS), 5.36, 5.51 (2s, 2H, CH₂=C), 7.15-7.40 (m with d at 7.29, $J=7.9$ Hz, 7H, Ph and 2H of *p*-Tol) and 7.71 (d, $J=7.9$ Hz, 2H, *p*-Tol); δ_{C} 21.66 (CH₃Ar), 34.35 (CH₂Ph), 48.17 (CH₂Cl), 68.16 (CHS), 122.22, 133.91 (CH₂=C), 126.88, 128.40, 129.10, 129.42, 129.65, 135.98, 136.48 and 145.07 (ArC); m/z 179 (M^+ -Ts, 18%), 157 (12), 144 (12), 143 (100), 142 (31), 141 (13), 129 (59), 128 (73), 127 (16), 115 (27), 91 (86), 77 (13) and 65 (30).

4-(Chloromethyl)-3-tosyl-4-pentenitrile (9f): ν (CHCl₃) 2240 (C≡N), 1300 and 1140 cm⁻¹ (SO₂); δ_{H} 2.46 (s, 3H, CH₃Ar), 2.92 (dd, $J=17.0$ and 10.5 Hz, 1H, CH₂CN), 3.08 (dd, $J=17.0$ and 5.0 Hz, 1H, CH₂CN), 4.08 (d, $J=5.7$ Hz, 2H, CH₂Cl), 4.13 (m, 1H, CHS), 5.39, 5.71 (2s, 2H, CH₂=C), 7.38 and 7.71 (2d, $J=8.2$ Hz, 4H, ArH); δ_{C} 18.23 (CH₂CN), 21.64 (CH₃Ar), 47.37 (CH₂Cl), 61.86 (CHS), 115.20 (CN), 123.42, 132.34 (CH₂=C), 129.28, 130.07, 135.49 and 146.22 (ArC); m/z 283 (M^+ , 0.5%), 248 (10), 155 (54), 128 (12), 91

(100), 65 (42) and 63 (10). High resolution mass spectrum required for $C_{13}H_{14}ClNO_2S$: m/z 283.0434. Found: 283.0424.

Ethyl 4-(Chloromethyl)-3-tosyl-4-pentenoate (9g): ν ($CHCl_3$) 1730 (C=O), 1300 and 1140 cm^{-1} (SO_2); δ_H 1.14 (t, $J=7.1$ Hz, 3H, CH_3CH_2), 2.38 (s, 3H, CH_3Ar), 2.72 (dd, $J=16.2$ and 10.1 Hz, 1H, CH_2CO), 3.02 (dd, $J=16.2$ and 5.0 Hz, 1H, CH_2CO), 4.00 (q, $J=7.1$ Hz, 2H, CH_2CH_3), 4.03 (dd, $J=12.8$ and 0.9 Hz, 1H, CH_2Cl), 4.16 (dd, $J=12.8$ and 1.1 Hz, 1H, CH_2Cl), 4.25 (dd, $J=10.1$ and 5.0 Hz, 1H, CHS), 5.13, 5.51 (2s, 2H, $CH_2=C$), 7.29 and 7.65 (2d, $J=8.1$ Hz, 4H, ArH); δ_C 13.90 (CH_3CH_2), 21.53 (CH_3Ar), 33.93 (CH_2CO), 47.03 (CH_2Cl), 61.21 (CH_2O), 63.43 (CHS), 121.91, 132.96 ($CH_2=C$), 129.36, 129.62, 137.10, 145.33 (ArC) and 168.91 (CO); m/z 331 (M^+ , 2%), 177 (32), 175 (100), 149 (10), 147 (29), 139 (22), 133 (25), 105 (16), 92 (13), 91 (39), 67 (33) and 65 (33). Anal. calcd. for $C_{15}H_{19}ClO_4S$: C, 54.46; H, 5.79; and S, 9.69. Found: C, 54.44; H, 5.78; and S, 10.01.

(E)-5,5-Dimethyl-2-methylene-3-tosyl-3-hexen-1-ol (15): ν 3500 (OH), 3080, 1640 ($CH=C$), 1300 and 1140 cm^{-1} (SO_2); δ_H 1.13 [s, 9H, (CH_3)₃C], 2.42 (s, 3H, CH_3Ar), 2.95 (br. s, 1H, OH), 4.25 (s, 2H, CH_2O), 4.45, 5.39 (2s, 2H, $CH_2=C$), 7.00 (s, 1H, $CH=C$), 7.28 and 7.66 (2d, $J=7.9$ Hz, 4H, ArH); δ_C 21.47 (CH_3Ar), 29.69 [(CH_3)₃C], 34.66 [$C(CH_3)_3$], 65.45 (CH_2O), 120.69, 138.81, 140.24, 151.40 ($CH_2=C$ and $CH=C$), 128.58, 129.31, 135.01 and 144.05 (ArC); m/z 294 (M^+ , 2%), 197 (10), 157 (11), 140 (20), 139 (40), 123 (25), 121 (17), 105 (21), 93 (30), 91 (92), 77 (35), 67 (20), 57 (100), 55 (24) and 41 (60). Anal. calcd. for $C_{16}H_{22}O_3S$: C, 65.27; H, 7.53; and S, 10.89. Found: C, 65.07; H, 7.49; and S, 10.98.

6-(Chloromethyl)-5-tosyl-6-hepten-2-one (20): R_f 0.65 (ether); ν 3080, 1650 ($CH=C$), 1710 (C=O), 1300 and 1150 cm^{-1} (SO_2); δ_H 1.90 (m, 1H, CH_2CS), 2.08 (s, 3H, CH_3CO), 2.25 (m, 1H, CH_2CS), 2.40 (s, 3H, CH_3Ar), 2.52 (t, $J=7.0$ Hz, 2H, CH_2CO), 3.89 (dd, $J=11.5$ and 4.0 Hz, 1H, CHS), 4.11, 4.16 (2d, $J=16.5$ Hz, 2H, CH_2Cl), 5.20, 5.59 (2s, 2H, $CH_2=C$), 7.35 and 7.77 (2d, $J=8.2$ Hz, 4H, ArH); δ_C 21.55 (CH_3Ar), 22.10 (CH_2CS), 29.76 (CH_3CO), 39.32 (CH_2CO), 48.87 (CH_2Cl), 65.25 (CHS), 122.13, 133.53 ($CH_2=C$), 129.24, 129.58, 136.71, 145.00 (ArC) and 206.70 (CO); m/z 209 ($M^+-C_4H_7ClO$, 9%), 159 (10), 91 (14), 81 (15), 65 (10) and 43 (100).

6-(Chloromethyl)-1,1,1,3,3,5-hexadeuterio-5-tosyl-6-hepten-2-one (21): R_f 0.65 (ether); ν 3080, 1650 ($CH=C$), 1710 (C=O), 1300 and 1150 cm^{-1} (SO_2); δ_H 1.90, 2.26 (2d, $J=12.0$ Hz, 2H, CH_2D), 2.45 (s, 3H, CH_3Ar), 4.10, 4.14 (2d, $J=15.5$ Hz, 2H, CH_2Cl), 5.20, 5.68 (2d, $J=0.8$ Hz, 2H, $CH_2=C$), 7.35 and 7.73 (2d, $J=8.2$ Hz, 4H, ArH); δ_C 21.50 (CH_3Ar), 21.95 (CH_2CS), 29.36 (sept, $J=19.0$ Hz, CD_3CO), 39.30 (q, $J=18.5$ Hz, CD_2CO), 48.76 (CH_2Cl), 64.95 (t, $J=20$ Hz, CDS), 122.09 ($CH_2=C$), 129.19, 129.54, 133.48, 136.62, 144.96 (ArC and $C=CH_2$) and 212.95 (CO); m/z 165 (M^+-Ts , 6%), 91 (13), 84 (15), 65 (17), 46 (100), 45 (69), 44 (19) and 43 (10).

Reduction of Compounds 5f, 6c and 8a with Sodium Amalgam. General Procedure.

To a suspension of anhydrous Na_2HPO_4 (251 mg, 1.75 mmol) and ca. 6% sodium amalgam (1.70g, 4.4 mmol) in dry methanol (5ml) was dropped at 0°C a solution of the corresponding sulfone (0.44 mmol) in methanol (1.5 ml). The reaction mixture was stirred at the temperature and for the time shown in table 4 and monitored

by TLC and GLC. Then the reaction was hydrolyzed with water and extracted with CH_2Cl_2 (2x10 ml) and the organic layer was dried over anhydrous Na_2SO_4 and evaporated in vacuo (15 torr) yielding crude pure compounds 22-24 and 26. Yields and physical data are included on the table 4, spectral and analytical data follow:

2-[2-(2-Methylenecyclopropyl)ethyl]-1,3-dioxolane (22): ν 3080 and 1640 cm^{-1} ($\text{CH}=\text{C}$); δ_{H} 1.05-1.95 (m, 7H, $\text{CH}_2\text{CH}_2\text{CHO}$ and $\text{CHCH}_2\text{C}=\text{C}$), 3.78 (m, 4H, $2\times\text{CH}_2\text{O}$), 4.81 (t, $J=4.5$ Hz, 1H, CHO), 5.26 and 5.33 (2m, 2H, $\text{CH}_2=\text{C}$); δ_{C} 9.20 ($\text{CH}_2\text{C}=\text{C}$), 15.14 ($\text{CHC}=\text{C}$), 27.30, 33.34 ($\text{CH}_2\text{CH}_2\text{CO}$), 64.64 ($2\times\text{CH}_2\text{O}$), 102.49 ($\text{CH}_2=\text{C}$), 104.02 (CHO) and 136.38 ($\text{C}=\text{CH}_2$); m/z 154 (M^+ , 1%), 153 (M^+-1 , 4), 99 (13), 81 (16), 79 (13), 73 (100), 45 (35) and 41 (10).

2-[2-(2-Methyl-1-cyclopropenyl)ethyl]-1,3-dioxolane (23): ν 3080 and 1640 cm^{-1} ($\text{CH}=\text{C}$); δ_{H} 0.72 (s, 2H, $\text{CH}_2\text{C}=\text{C}$), 1.25-1.95 (m, 4H, $\text{CH}_2\text{CH}_2\text{CHO}$), 1.96 (s, 3H, CH_3), 3.89 (m, 4H, $2\times\text{CH}_2\text{O}$) and 4.84 (t, $J=4.5$ Hz, 1H, CHO); δ_{C} 8.04 ($\text{CH}_2\text{C}=\text{C}$), 11.18 (CH_3), 20.67, 31.59 ($\text{CH}_2\text{CH}_2\text{CO}$), 64.71 ($2\times\text{CH}_2\text{O}$), 103.75 (CHO) 105.91 and 108.92 ($\text{C}=\text{C}$); m/z 154 (M^+ , 1%), 153 (M^+-1 , 4), 99 (11), 81 (16), 79 (13), 73 (100), 45 (35) and 41 (10).

2-Isopropyl-4-methyl-2,5-dihydrofuran (24):¹⁸ ν 3080 and 1580 cm^{-1} ($\text{CH}=\text{C}$); δ_{H} 0.8, 0.85 [2t, $J=6.8$ Hz, 6H, (CH_3)₂CH], 1.62-1.81 (m with s at 1.71, 4H, CH_3CCH_2 and CHCH_3), 4.43 (deform. s, 2H, CH_2O), 4.54 (m, 1H, CHO) and 5.34 (deform. s, 1H, $\text{CH}=\text{C}$); δ_{C} 12.26 ($\text{CH}_3\text{C}=\text{C}$), 17.85 [(CH_3)₂CH], 33.59 (CHCH_3), 78.04 (CH_2O), 91.97 (CHO), 121.79 and 126.72 ($\text{C}=\text{C}$); m/z 126 (M^+ , 2%), 83 (100), 82 (12), 5 (17) and 43 (5).

erythro/threo-4-(1-Hydroxymethyl)-1-methylcyclopentene (26): ν 3360 cm^{-1} (OH); δ_{H} 1.09, 1.10 (2d, $J=7.0$ Hz, 3H, CH_3CO), 1.63 (s, 3H, $\text{CH}_3\text{C}=\text{C}$), 1.95-2.60 (m, 6H, $2\times\text{CH}_2\text{C}=\text{C}$, CHCO and OH), 4.11 (m, 1H, CHO) and 5.17 (broad s, 1H, $\text{CH}=\text{C}$); δ_{C} 16.51, 16.57 (CH_3CO), 21.49, 21.52 ($\text{CH}_3\text{C}=\text{C}$), 35.02, 35.62, 39.14, 39.73 ($2\times\text{CH}_2\text{C}=\text{C}$), 45.95, 45.97 (CHCO), 71.83, 71.85 (CO), 123.25, 123.51 ($\text{CH}=\text{C}$), 139.41 and 139.70 ($\text{C}=\text{CH}$); m/z 126 (M^+ , 4%), 108 (9), 93 (100), 91 (20), 82 (10), 79 (30), 77 (20), 67 (17), 53 (14), 45 (18), 43 (16) and 41 (13).

Reduction of Compound 7d with Samarium(II) Iodide. Synthesis of Compound 25.

Over a freshly prepared 0.2M solution of SmI_2^{23} (25 ml) was added, at -20°C , first the sulfone 7d (328 mg, 1.00 mmol) and then HMPA (4 ml), staying the mixture at -20°C during 1.5 hours. The reaction was hydrolyzed with 3M hydrochloric acid (20 ml) and extracted with ether (2x10 ml). The ethereal solution was dried over anhydrous Na_2SO_4 and evaporated *in vacuo* (15 torr). The residue was purified by column chromatography (hexane/ether) yielding the pure compound 25 (see table 4).

4-Methyl-2-phenyl-4-penten-2-ol (25): ν 3450 (OH), 3080 and 1640 cm^{-1} ($\text{CH}=\text{C}$); δ_{H} 1.32 (s, 3H, CH_3CO), 1.49 (s, 3H, $\text{CH}_3\text{C}=\text{C}$), 2.20 (broad s, 1H, OH), 2.44, 2.57 (2d, $J=13.0$ Hz, 2H, CH_2), 4.67, 4.82 (2s, 2H, $\text{CH}_2=\text{C}$), 7.15-7.41 (m, 5H, Ph); δ_{C} 24.18 ($\text{CH}_3\text{C}=\text{C}$), 30.66 (CH_3CO), 52.02 (CH_2CO), 76.58 (CO), 115.64 ($\text{CH}_2=\text{C}$), 124.70, 124.77, 126.48, 128.05 and 142.56 ($\text{C}=\text{CH}_2$ and ArC); m/z 174 (M^+-2 , 1%), 159 (15), 128 (26), 121 (87), 105 (19), 77 (29), 51 (15) and 43 (100).

ACKNOWLEDGMENTS

We thank DGICYT (Projects PB88-0287 and PB91-0751) from the Spanish Ministerio de Educación y Ciencia (MEC) for financial support. J. M. S. thanks also MEC for a predoctoral fellowship.

REFERENCES AND NOTES

1. (a) Schank, K. in *Methoden der Organischen Chemie (Houben-Weyl)*; Georg Thieme Verlag: Stuttgart, 1985, Vol. E/11. (b) *The Chemistry of Sulphones and Sulphoxides*; Patai, S.; Rappoport, Z.; Stirling, C. Eds.; J. Wiley & Sons: Chichester, 1988. (c) Simpkins, N. S. in *Sulphones in Organic Synthesis*; Pergamon Press: Oxford, 1993.
2. (a) Volhardt, J.; Gais, H.-J.; Lukas, K. L. *Angew. Chem. Int. Ed. Engl.* **1985**, *24*, 608-610. (b) Gais, H.-J.; Ball, W. A.; Bund, J. *Tetrahedron Lett.* **1988**, *29*, 781-784. (c) Gais, H.-J.; Volhardt, J. *Tetrahedron Lett.* **1988**, *29*, 1529-1532.
3. For preparation and other synthetic applications of this reagent see: (a) Nájera, C.; Sansano, J. M. *Tetrahedron* **1992**, *48*, 5179-5190. (b) Nájera, C.; Sansano, J. M. *Tetrahedron Lett.* **1993**, *34*, 3781-3784.
4. Preliminary communication: Nájera, C.; Sansano, J. M. *Tetrahedron Lett.* **1992**, *33*, 6543-6546.
5. Monolithium compounds derived from 2-(chloromethyl)-3-tosylpropene^{3b} or from the phenylsulfonyl homologue⁶ have been already prepared by reaction with one equivalent of n-butyllithium at low temperatures.
6. Breuilles, P.; Uguen, D. *Tetrahedron Lett.* **1988**, *29*, 201-204.
7. In absence of DMPU the monolithium^{3b} derivative of compound **1** is formed.
8. Arylsulfonylated methylenecyclopropanes are adequate precursors for methylenecyclopropenes: (a) Billups, W. F.; Blakeney, A. J.; Chamberlain, W. T. *J. Org. Chem.* **1976**, *41*, 3771-3772. (b) Roberts, S. W.; Stirling, C. J. M. *J. Chem. Soc., Chem. Commun.* **1991**, 170-171.
9. For studies about intramolecular nucleophilic substitution of sulfonyl carbanions bearing a ω -leaving group, see: Jeffery, S. M.; Stirling, C. J. M. *J. Chem. Soc., Perkin Trans. 2* **1993**, 1617-1624.
10. This kind of dihydrofurans has been prepared by reaction of lithium (*E*)-3-lithio-2-methyl-3-tosylpropenolate with aldehydes followed by intramolecular etherification: Nájera, C.; Yus, M. *J. Org. Chem.* **1989**, *54*, 1491-1499.
11. Truce, W. E.; Klinger, T. C. *J. Org. Chem.* **1970**, *35*, 1834-1838.
12. For a review about functionalized organolithium compounds see: Nájera, C.; Yus, M. *Trends in Organic Chemistry* **1991**, *2*, 155-181.
13. For synthesis of 2-sulfonyl-1,3-dienes starting from dienes and synthetic applications, see, for instance: (a) Andell, O. S.; Bäckvall, J. E. *Tetrahedron Lett.* **1985**, *26*, 4555-4558. (b) Bäckvall, J. E.; Nájera, C.; Yus, M. *Tetrahedron Lett.* **1988**, *29*, 1455-1458. (c) Bäckvall, J. E.; Juntunen, S. K. *J. Am. Chem. Soc.* **1987**, *109*, 6396-6403. (d) Bäckvall, J. E.; Ericsson, A. M.; Juntunen, S. K.; Nájera, C.; Yus, M. *J. Org. Chem.* **1993**, *58*, 5211-5225.
14. Cuvigny, T.; Hervé du Penhoat, C.; Julia, M. *Tetrahedron* **1986**, *42*, 5329-5336.
15. Similar reaction has been carried out with the monolithium derivative of 2-(bromomethyl)-3-(phenylsulfonyl)-1-propene and electrophilic olefins to give sulfonylated methylenecyclopentenes¹⁶.

16. Ghera, E.; Yechezkel, T.; Hassner, A. *Tetrahedron Lett.* **1990**, *31*, 3653-3656.
17. Trost, B. M.; Arndt, H. C.; Strege, P. E.; Verhoeven, T. R. *Tetrahedron Lett.* **1976**, 3477-3478.
18. Ibragimov, I. I. *Uses. Nauchn. Konf. Khim. Teknol. Furanovykh Soedin [Tezisy Dokl.]* **1978**, 6-8; *Chem. Abstr.* **1980**, *93*, 46270m.
19. This type of methylenecyclopropanes are related to hypoglycine, see, for instance: Achmatowicz, B.; Kabat, M. M.; Krajewski, J.; Wicha, J. *Tetrahedron* **1992**, *48*, 10201-10210.
20. Künzer, H.; Stankhe, M.; Sauer, G.; Weichert, R. *Tetrahedron Lett.* **1991**, *32*, 1949-1952.
21. (a) Bremmer, J.; Julia, M.; Launay, M.; Stacino, J-P. *Tetrahedron Lett.* **1982**, *23*, 3265-3266. (b) Lauron, H.; Stacino, J-P.; Verpeaux, J-N.; Jeannin, Y.; Dromzee, Y. *Tetrahedron* **1986**, *42*, 2475-2484.
22. In the synthesis of compounds **3**, **12b**, and **20-21**, the reaction mixture was maintained 20 min at -90°C before hydrolysis.
23. Girard, P.; Namy, J. L.; Kagan, H. B. *J. Am. Chem. Soc.* **1980**, *102*, 2693-2698.

(Received in UK 15 November 1993; revised 20 December 1993; accepted 7 January 1994)